

AMENDMENTS TO THE SPECIFICATION
(other than claims)

Please replace paragraph beginning at page 1, line 15, which starts with "Because diamond is composed of," with the following amended paragraph:

~~Because d~~Diamond is composed of ~~silicon (Si), which is widely used in semiconductor materials, and carbon (C), which is a Group IVb element in the same family as silicon (Si), which is widely used in semiconductor materials,~~ and because ~~it~~ diamond possesses the same crystalline structure as Si, diamond may be regarded as a semiconductor material. In terms of being a semiconductor material diamond has an extraordinarily large bandgap of 5.5 eV, and a high carrier mobility of $2000 \text{ cm}^2/\text{V} \cdot \text{s}$ for electrons/holes alike at room temperature. And with its dielectric constant being a small 5.7, its breakdown electric field is a large 5×10^6 V/cm. Diamond also has the unusual property of negative electron affinity, in that its vacuum level is present below the lower edge of its conduction band.